

Updates to the Cramer et al. Decision Tree and the Threshold of Toxicological Concern: the Expanded Decision Tree and Its Use in Safety Assessments

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November 10, 2021

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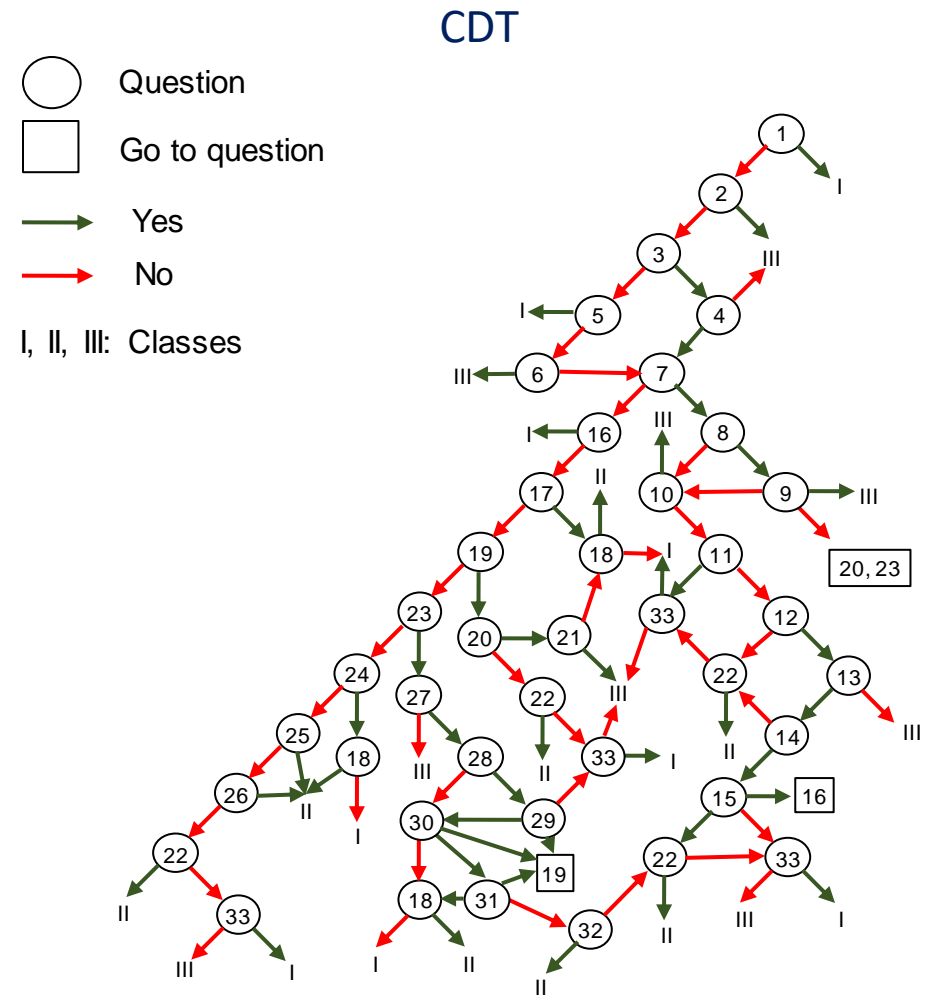
- Our goals
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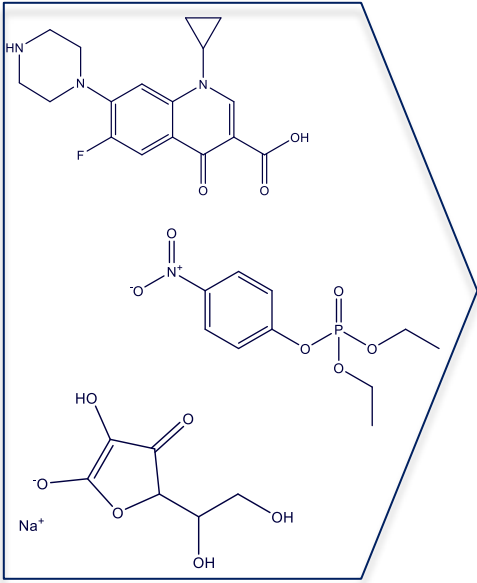
The Cramer et al. Decision Tree (CDT) (1978)

The CDT is a sequence of **yes/no** questions that either leads to another question, or the assignment of the substance to one of three classes of relative chronic oral toxicity.

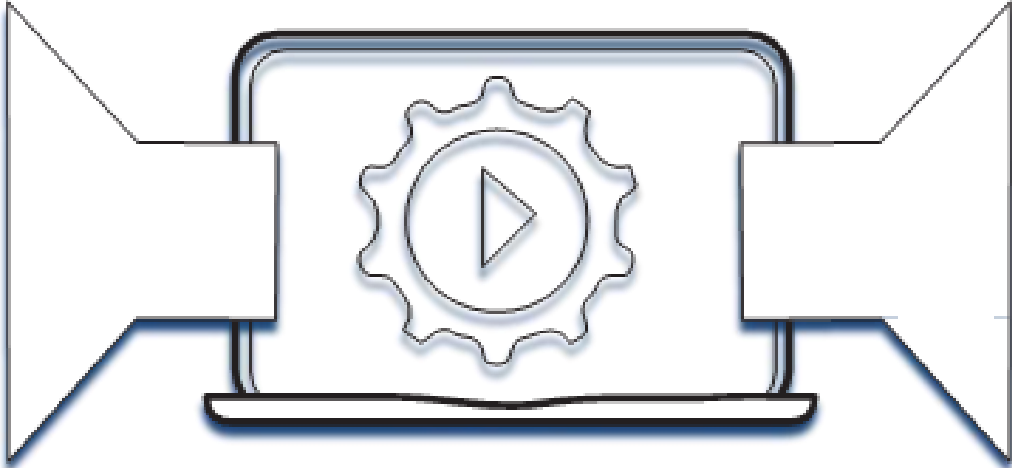
- 33 main questions
- 11 sub-questions



Three classes of relative chronic oral toxic potential



CDT



State-of-the-art at the time of its publication (1978)

Relative Chronic Oral Toxicity

Class I: Low order of oral toxicity

Class II: Intermediate Toxicity

Class III: Substances having structural features that suggest either “significant toxicity” or “permit no strong initial presumptions of safety”

The Threshold of Toxicological Concern (TTC)



- 1996: a database of 613 substances was compiled and each substance was assigned to its CDT class. For each CDT class, a Threshold of Toxicological Concern (TTC) value was calculated (Munro *et al.*, 1996).

$$\text{Class TTC } [\mu\text{g/p/day}] = \frac{5^{\text{th}} \text{ percentile NEL [mg/kg bw/day]} \times \text{Avg. bw (60 kg bw/p)} \times 1000 \mu\text{g/mg}}{100 \text{ (safety factor)}}$$

	Class I	Class II	Class III
TTC ($\mu\text{g/p/day}$)	1,800	540	90

- If intake \leq class TTC: reasonably expected to be safe.



Why and how were the
CDT and its associated
TTCs updated?

Limitations of the CDT, Munro DB & TTC



CDT was designed with mostly flavors in mind: limited scope

Munro Database (DB) is relatively small (613)

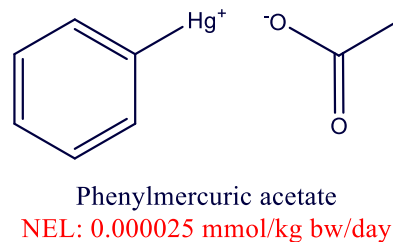
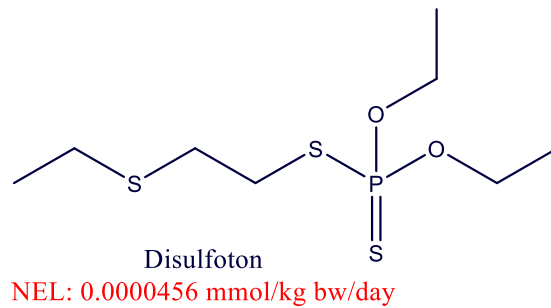
Class II TTC is based on NELs for only 28 compounds

Update: Expanding the size of the database

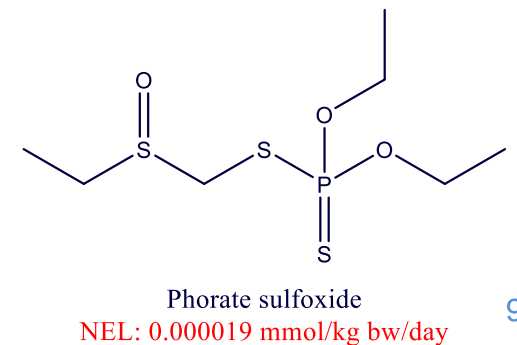
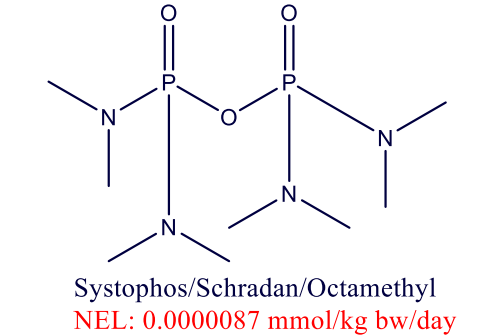
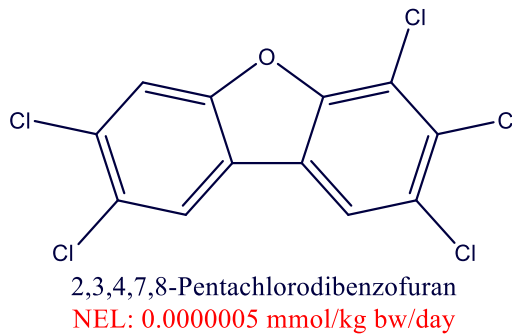
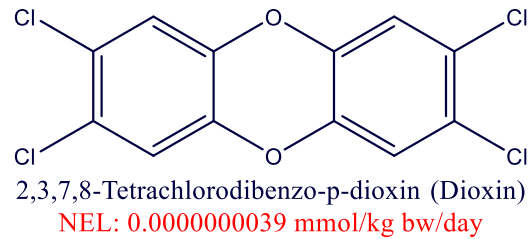


		Non-toxic	Low toxicity	Medium toxicity	High toxicity	Very high toxicity	Extreme toxicity
Classes		I	II	III	IV	V	VI
No. of compounds	EDT DB (used for calc.)	174	240	242	393	173	41
	EDT DB (total)*	219	349	344	613	334	50
	Classes	I		II	III		
	Munro, 1996, DB	137		28	448		

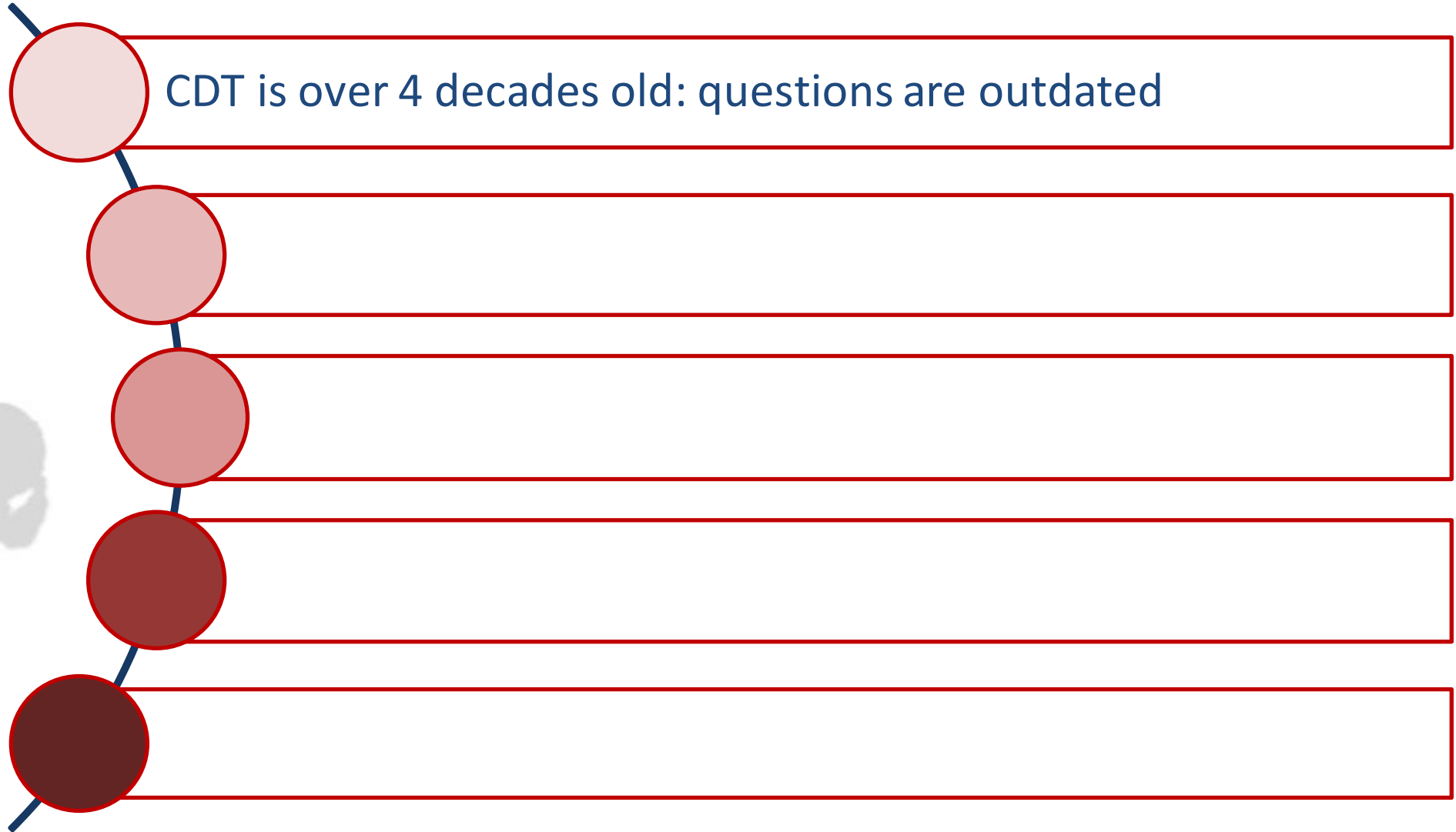
* Some substances in the EDT have only a LEL, hence they were not used to calculate class TTCs, but were used to help formulate EDT questions.



NELs are adjusted for duration

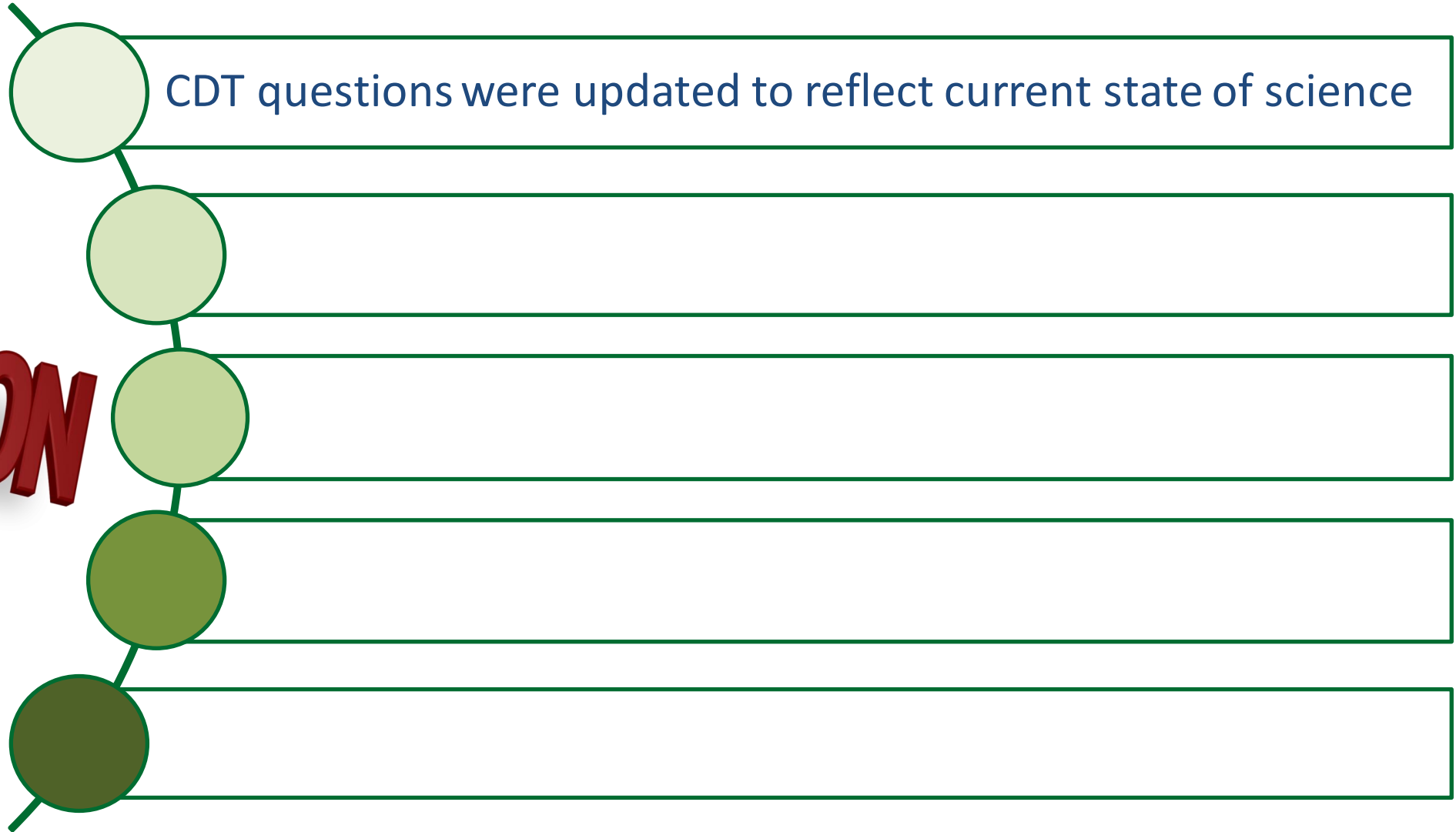


Additional limitations of the CDT



CDT is over 4 decades old: questions are outdated

Solutions for the limitations of the CDT



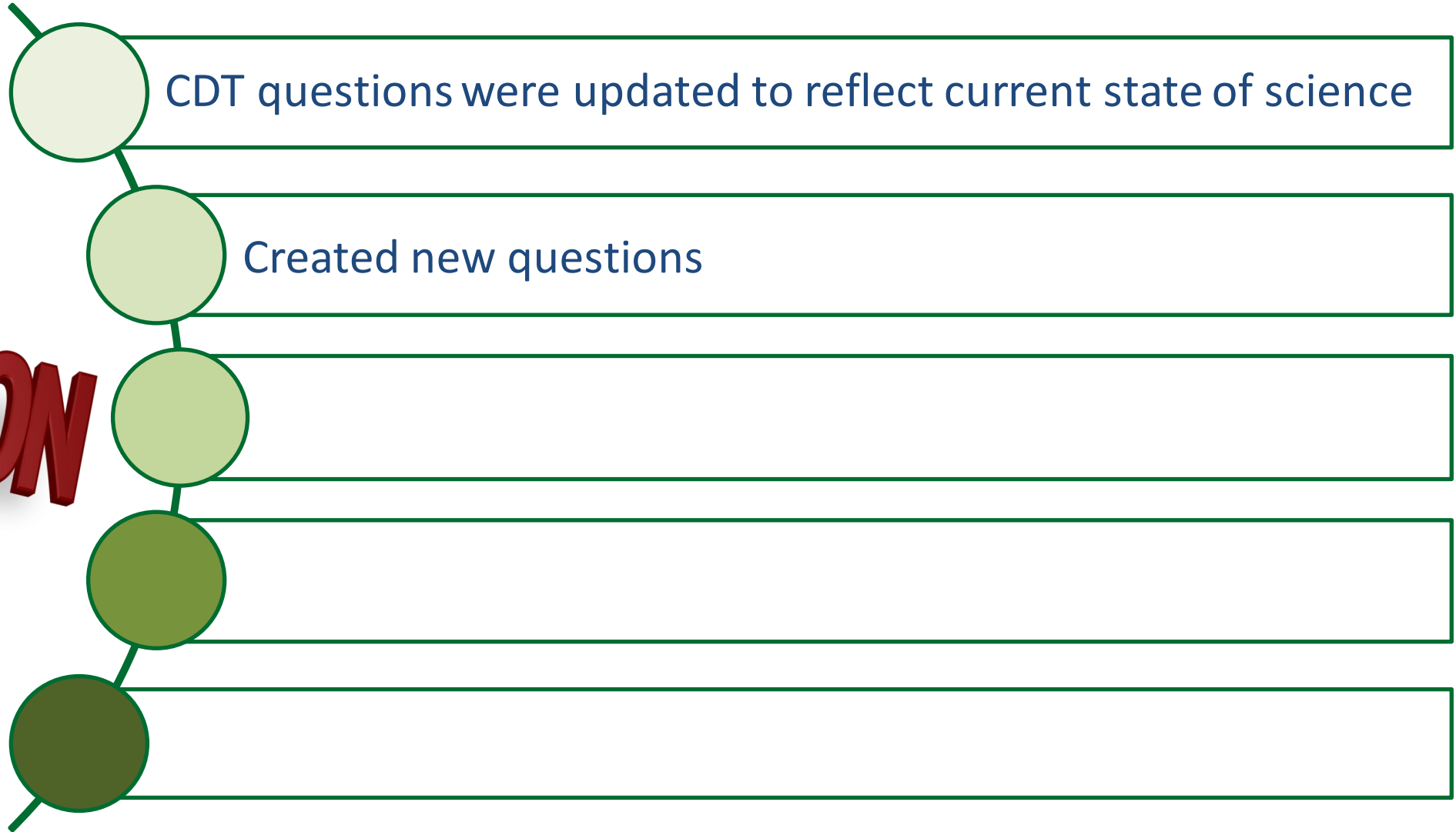
CDT questions were updated to reflect current state of science

Additional limitations of the CDT



- CDT is over 4 decades old: questions are outdated
- Limited number of elements, moieties, functional groups, and congeneric groups
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Solutions for the limitations of the CDT

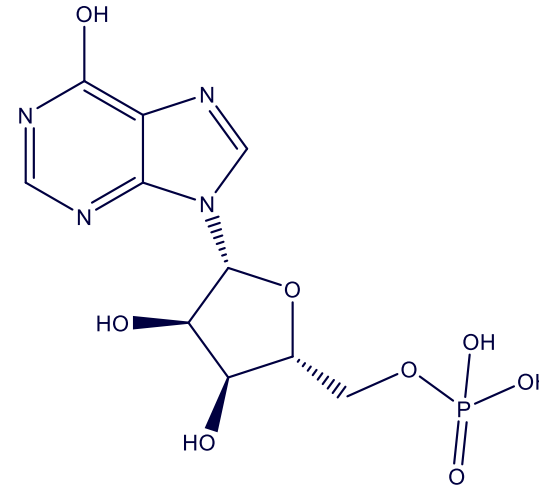




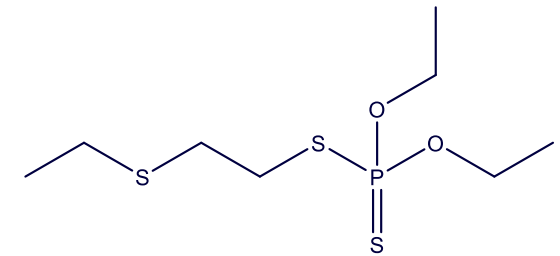
Examples of new questions created to address additional elements, functional groups, reactive moieties, and congeneric groups of compounds

Example 1: Covalently bound phosphorus (new element)

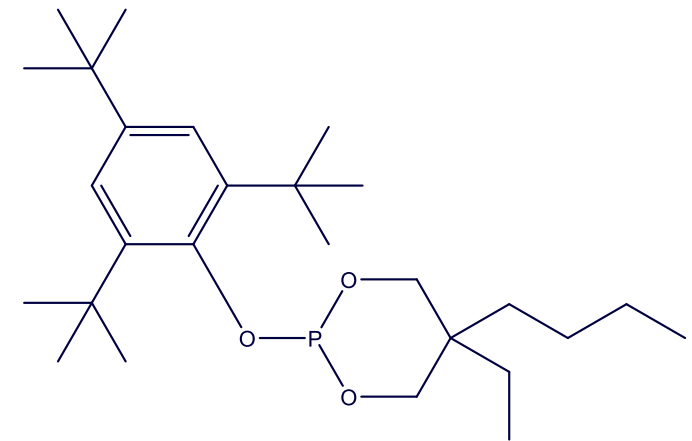
- **P** is found in a wide variety of compounds used in food contact materials, plastics in general, pharmaceuticals, solvents, herbicides, rodenticides, insecticides, some natural and synthetic toxins.
- **EDT DB** has 82 organophosphorus compounds and 20 compounds containing both P and one or more halogens. They exhibit a very broad range of toxicities.
- **CDT** defaults substances containing covalently bound P into CDT Class III unless they are a “normal constituent of the body”.



Inosine monophosphate NEL:
 2000 mg/kg bw/day or
 5.744 mmol/kg bw/day
 Widely used as a flavor enhancer



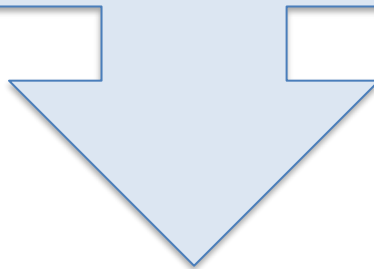
Disulfoton NEL:
 0.0125 mg/kg bw/day or
 0.0000456 mmol/kg bw/day
 Insecticide



Ultrinox 641 NEL:
 2.9 mg/kg bw/day or 0.015 mg/kg bw/day
 Antioxidant/stabiliser in polymers

Example 1: Phosphorus (new element)

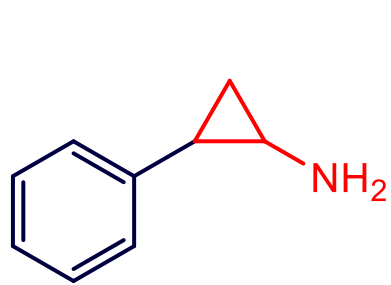
- **Q1i:** Nucleotides, nucleosides, phospholipids, monophosphates of amino acids, or their hydrolysis products (Class I)
- **Q2a,b,c,d:** classify a wide variety of P-containing substances into Classes III, V, and VI
- **Q6b,c,d:** capture very highly toxic compounds that may (or may not) contain P (Class V)



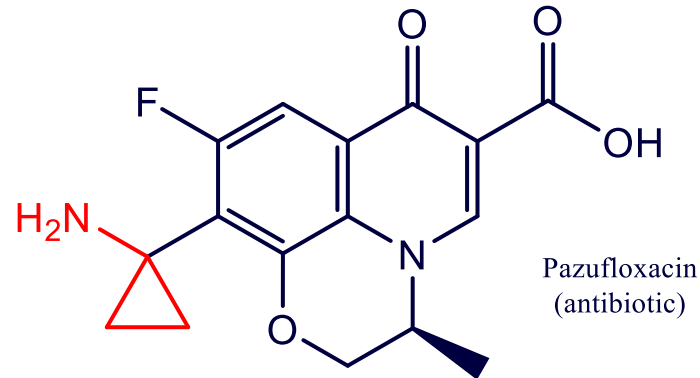
Compounds containing covalently bound P can be sorted into EDT Classes I, III, V, and VI to better differentiate their true toxic potentials.

Example 2: Aminocyclopropyl moiety

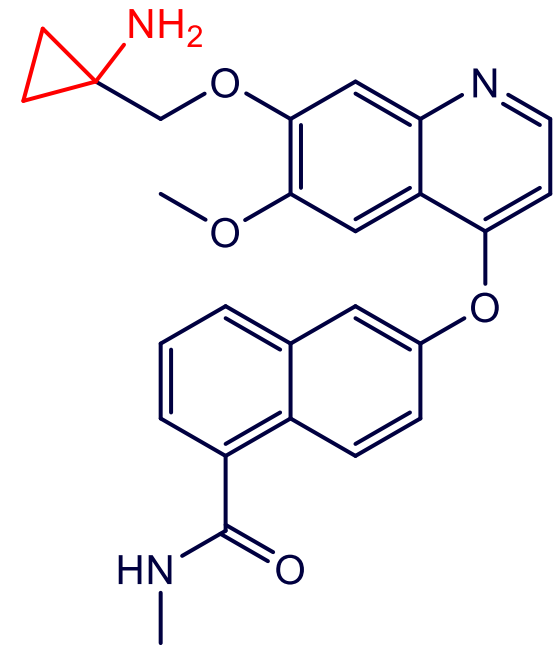
Cyclopropylamines are prevalent motifs in pharmaceuticals and agrochemicals:



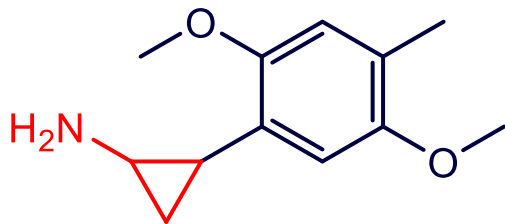
Tranylcypromine (antidepressant, monoamine oxidase inhibitor)



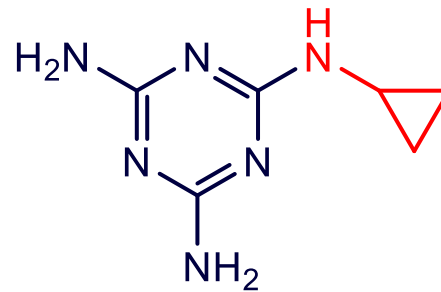
Pazufloxacin (antibiotic)



Lucitanib (investigational angiogenesis inhibitor)

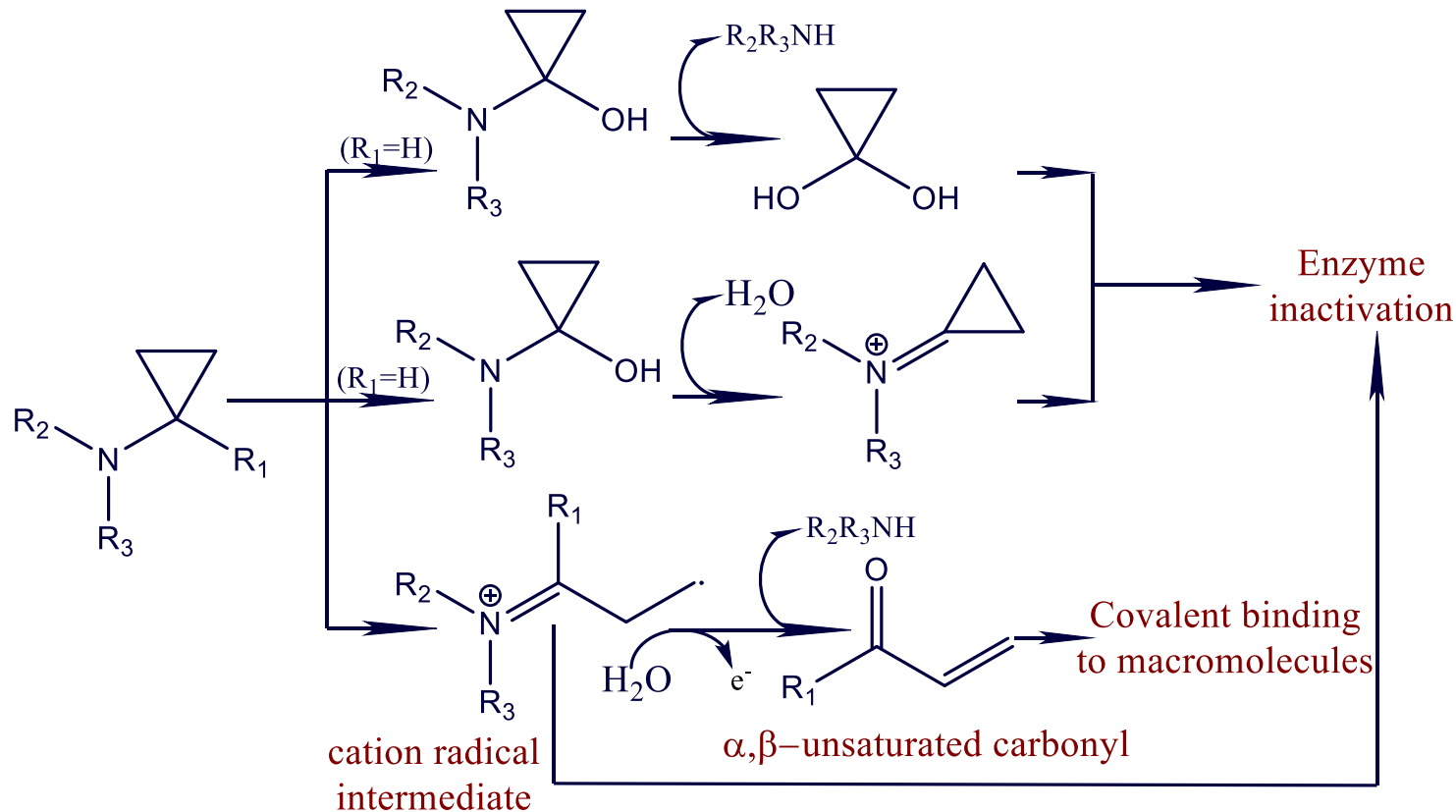


2,5-Dimethoxy-4-methylphenylcyclopropylamine (DMCPA, psychedelic drug)



Cyromazine (insecticide and an acaricide)

Example 2: Aminocyclopropyl moiety



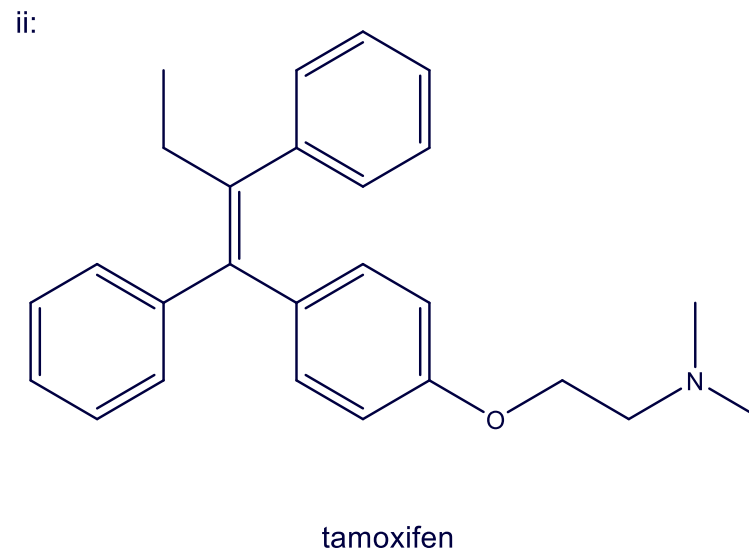
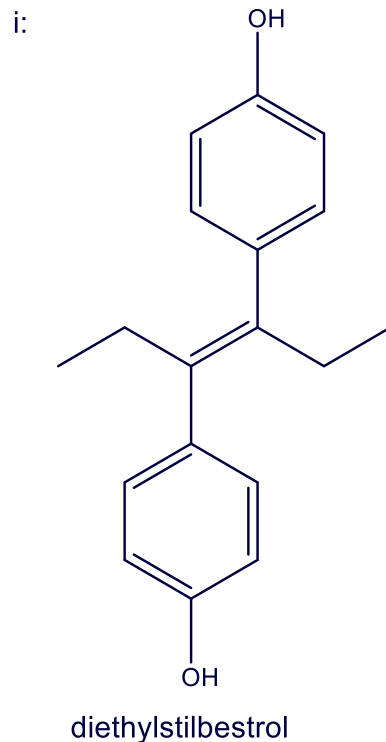
The cyclopropylamine moiety is a known inactivator of CYP450 and other enzymes such as MAO and ADH.

Q28r: Does the substance contain at least one aminocyclopropyl moiety?
If yes, assign to Class V.

Example 3: Selected nonsteroidal estrogens (new congeneric groups)



Q6g aims to capture nonsteroidal estrogens, namely **stilbestrols** (Q6gi) & **triphenylethylenes** (Q6gii). As their name suggest, they are selective estrogen receptor modulators.



Example 3: Selected nonsteroidal estrogens (new congeneric groups)



Q6g i) two benzene rings connected by a 2- or 3-carbon chain (connector, with or without unsaturation) and a hydroxy, corresponding ester, methoxy, and/or ether in the para position on each ring with or without methyl, ethyl, and/or ethylidene substitution on one or more connector carbons (not more than one per carbon). **One or more halogen(s) are allowed anywhere on the molecule along with methyl group(s) in the meta position on the benzene ring(s)** **or ii)** two benzene rings connected by a -C=C- (connector) and one connector carbon is substituted by a benzene ring (a total of three benzene rings) and the other connector carbon is either unsubstituted or substituted by a methyl or ethyl group or a halogen. **Any or all of the benzene rings may be substituted by a hydroxy, corresponding ester, methoxy, and/or ether in the para position, but this is not required.**

Additional limitations of the CDT



- CDT is over 4 decades old: questions are outdated
- Limited number of elements, moieties, & functional groups
- Ambiguous phrases and definitions
-
-

Elimination of ambiguous phrases & definitions



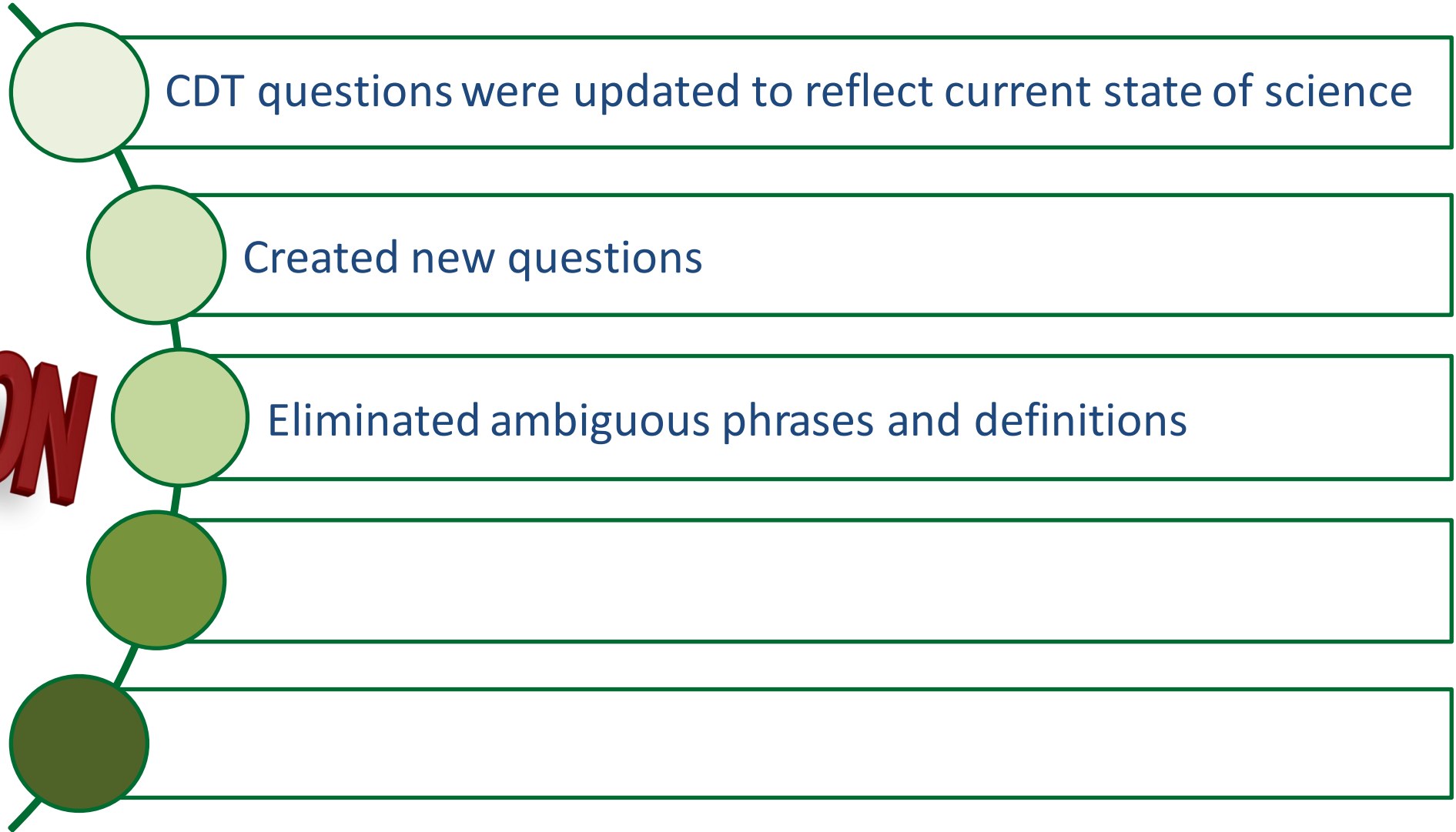
Common terpene: an isoprenoid compound reported in the literature as a more than trace constituent of two or more generally consumed foods, either raw or as ordinarily prepared for consumption, without added ingredients.

Sterically hindered: means posing steric hindrance to a functional group equivalent to or greater than that exhibited by *o*-tert-butyl or a 2,6-disubstitution on an aromatic ring.

Normal constituent of the body:.....

Common component of food:.....

Solutions for the limitations of the CDT



Additional limitations of the CDT



- CDT is over 4 decades old: questions are outdated
- Limited number of elements, moieties, & functional groups
- Ambiguous phrases and definitions
- Non-structure-based questions
-

Elimination of non-structure-based and/or ambiguous questions and phrases



Q1: Is the substance a *normal constituent of the body* or an optical isomer of such? (Class I)

Q22: Is the substance a *common component of food* or *structurally closely related* to a common component of food? (Class II)

Solutions for the limitations of the CDT

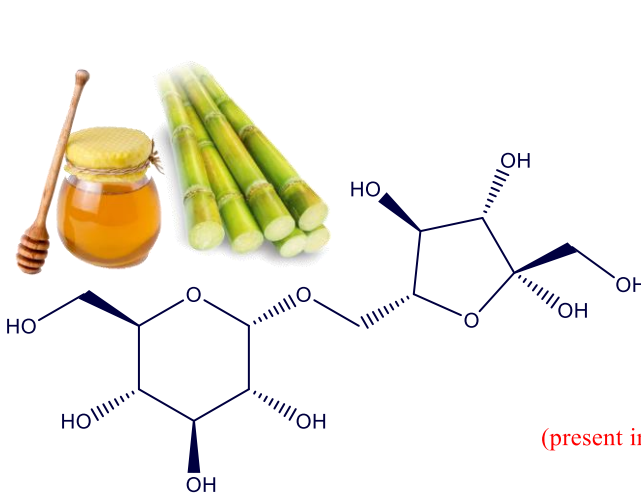


- CDT questions were updated to reflect current state of science
- Created new questions
- Eliminated ambiguous phrases and definitions
- Eliminated non-structure-based questions
-

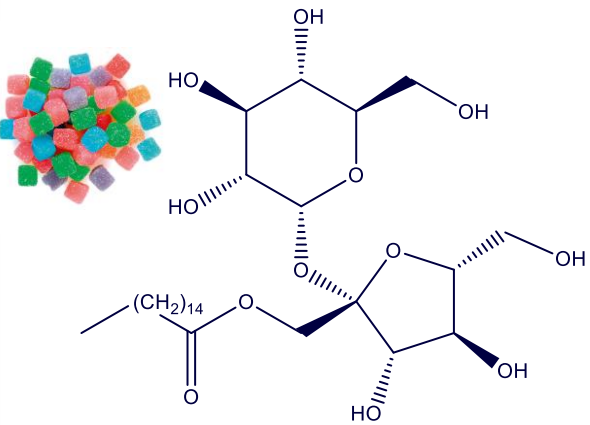
Replacing non-structure-based questions with structure-based questions (e.g. Q1a, b, c, d, e, f, g, h, i, j, & k)

Q1e) Monosaccharides and hydrolysable oligosaccharides and polysaccharides in addition to simple monosaccharide derivatives (phosphate esters, deoxy sugars, amino sugars, mono- and poly-methylated, sulfated, and sulfonic acid derivatives of monosaccharides.....)

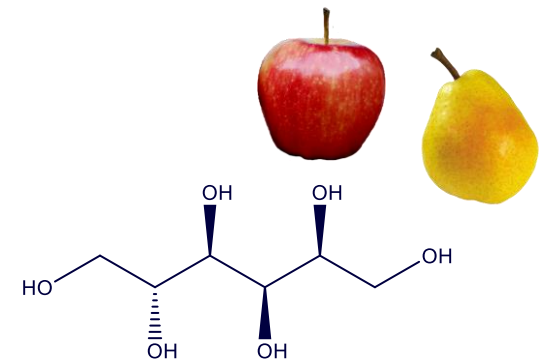
Q1f) Sugar alcohols or sugar acids and their corresponding esters in addition to derivatives of sugar alcohols that are both alkoxyated and esterified...



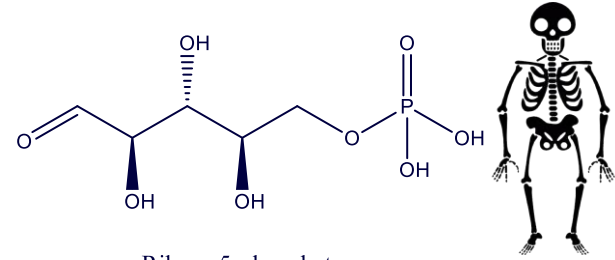
Isomaltulose
(present in honey and sugarcane extracts, used to produce isomalt (a sugar alcohol))



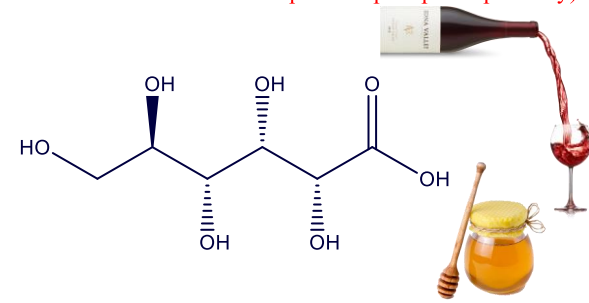
Sucrose monopalmitate (sucrose esters: emulsifiers)



Sorbitol
(present in apples, pears, peaches, and prunes or made from potato starch)



Ribose 5-phosphate
(both a product and an intermediate of the pentose phosphate pathway)



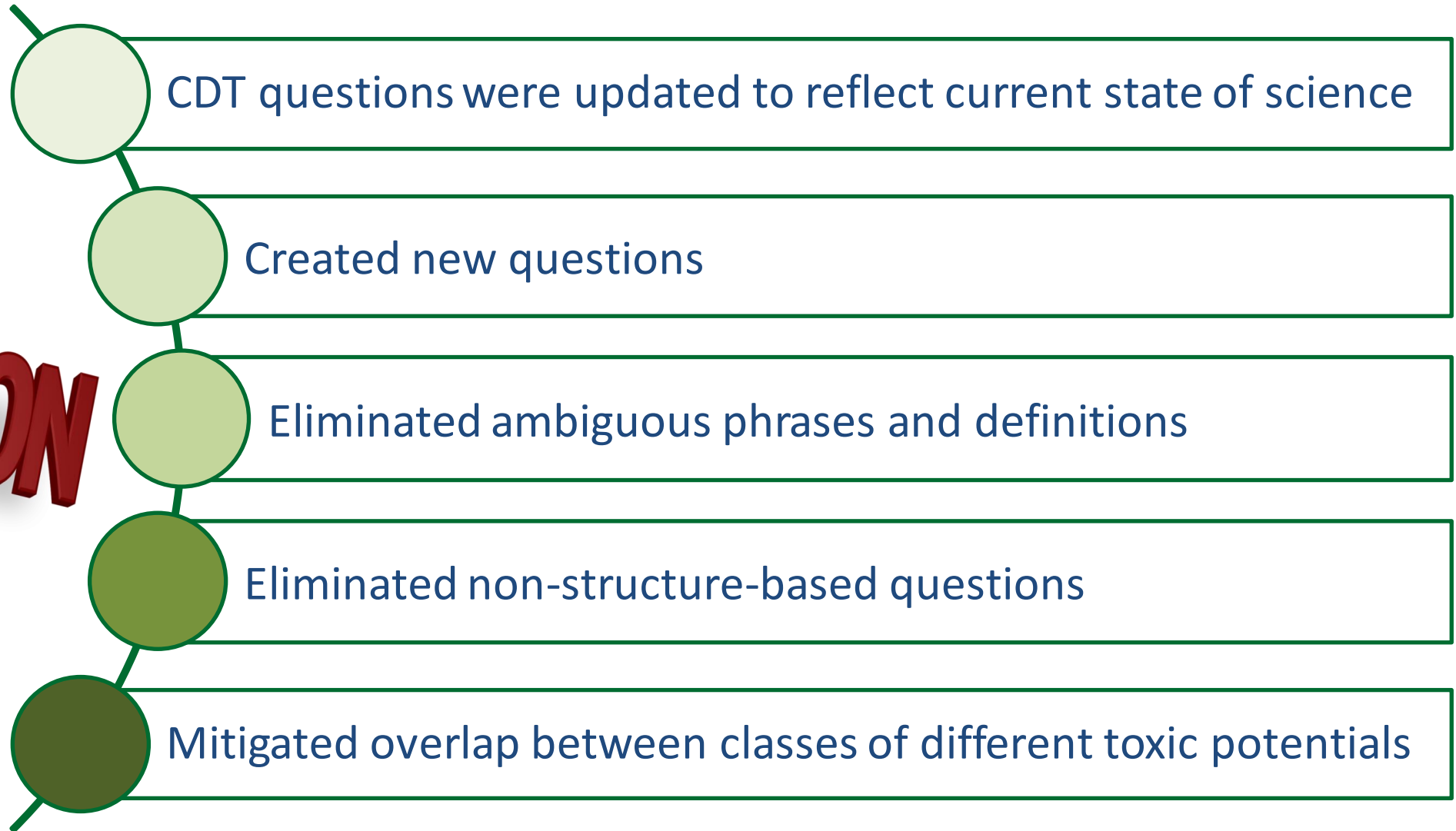
Gluconic acid (sugar acid)
(found in fruit, honey, and wine; food additive (acidity regulator))

Additional limitations of the CDT



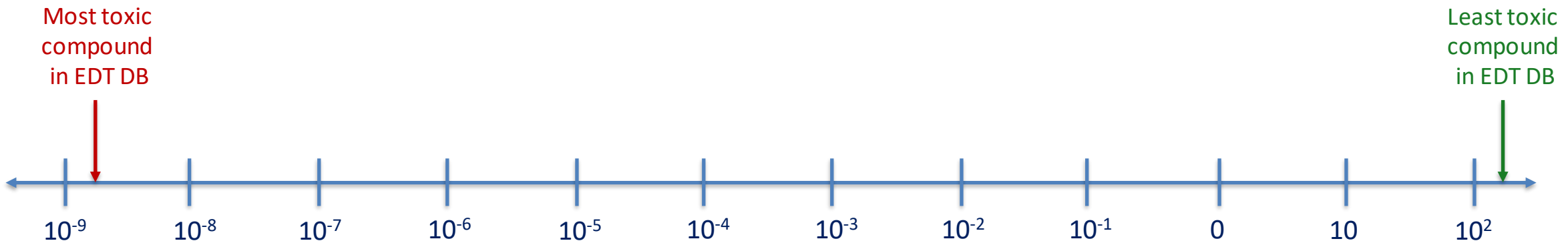
- CDT is over 4 decades old: questions are outdated
- Limited number of elements, moieties, & functional groups
- Ambiguous phrases and definitions
- Non-structure-based questions
- Large overlap between classes of different toxic potentials

Solutions for the limitations of the CDT

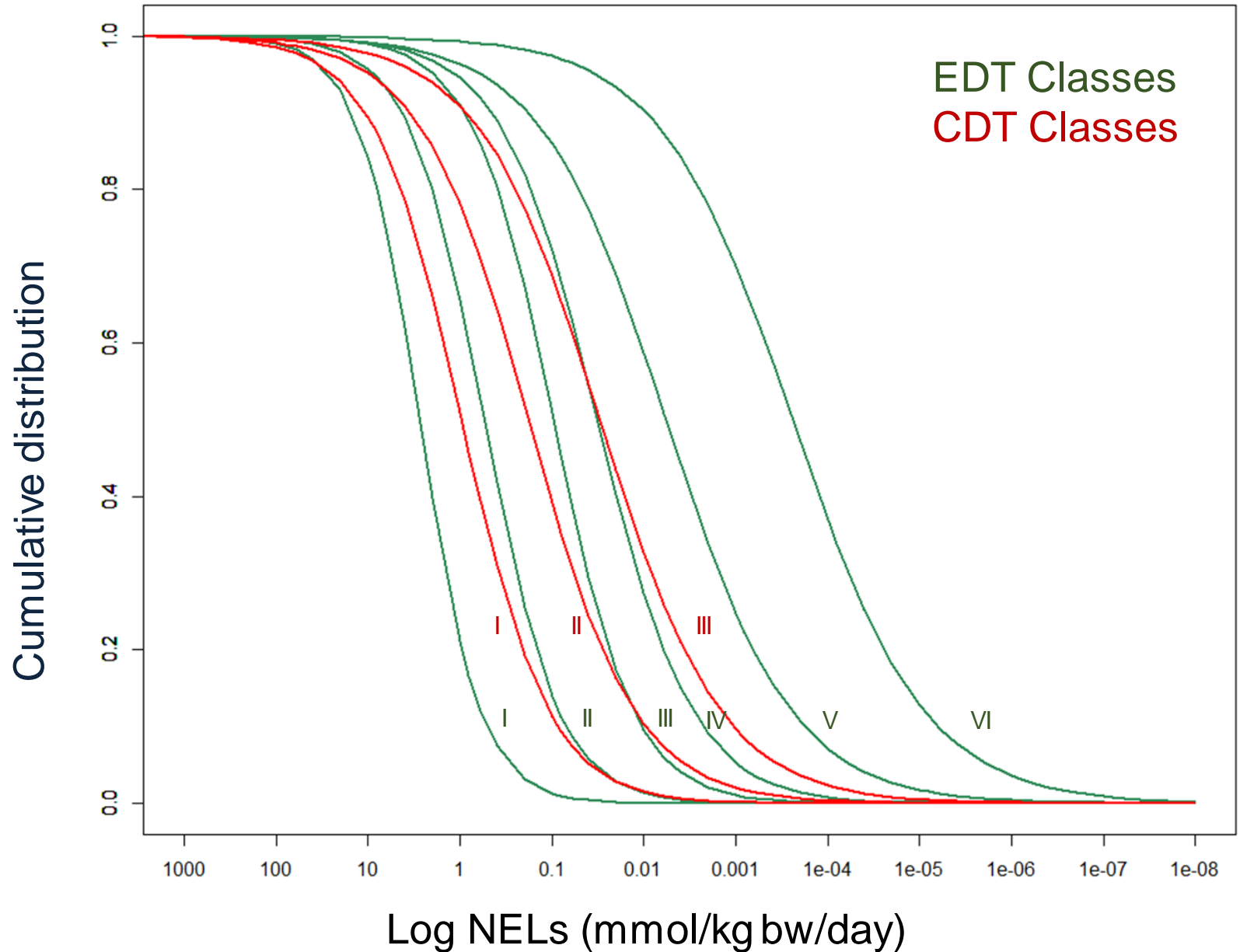




Why were the number of classes doubled?



NEL range [mmol/kg bw/day] in the EDT DB used for TTC calculations



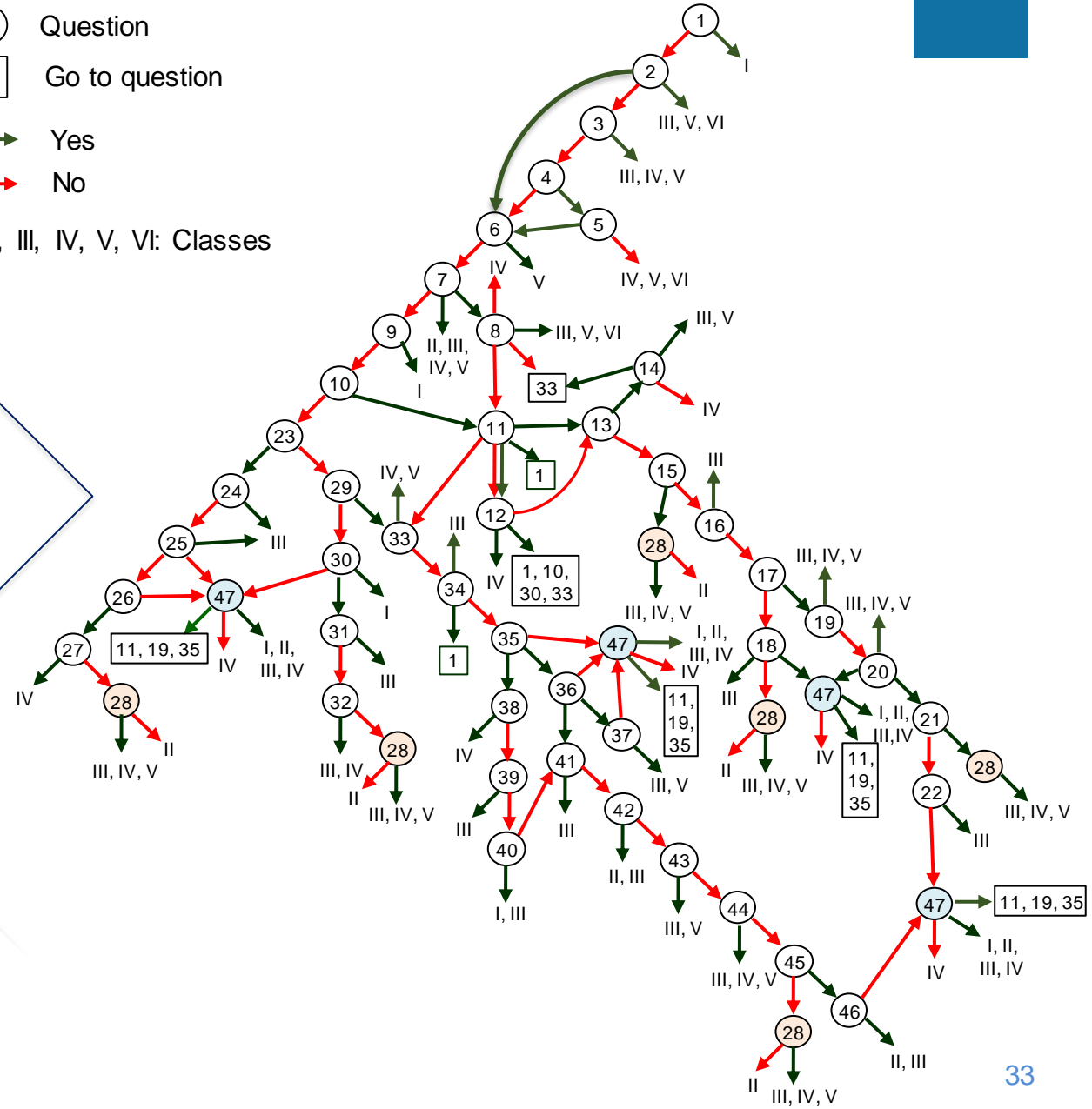
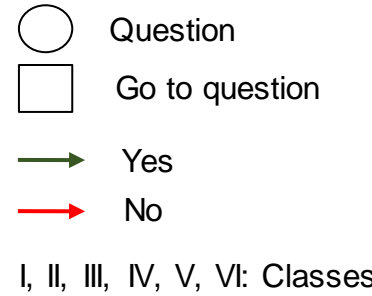
Comparison of NEL distributions for compounds in the EDT DB grouped into their CDT and EDT classes:
significantly decreased overlap with the EDT



Results: the EDT and
the updated TTCs

The Expanded Decision Tree (EDT)

The EDT is a sequence of (chemical) structure-based **yes/no** questions that either leads to another question, or the assignment of the substance to one of six classes of relative chronic oral toxicity.



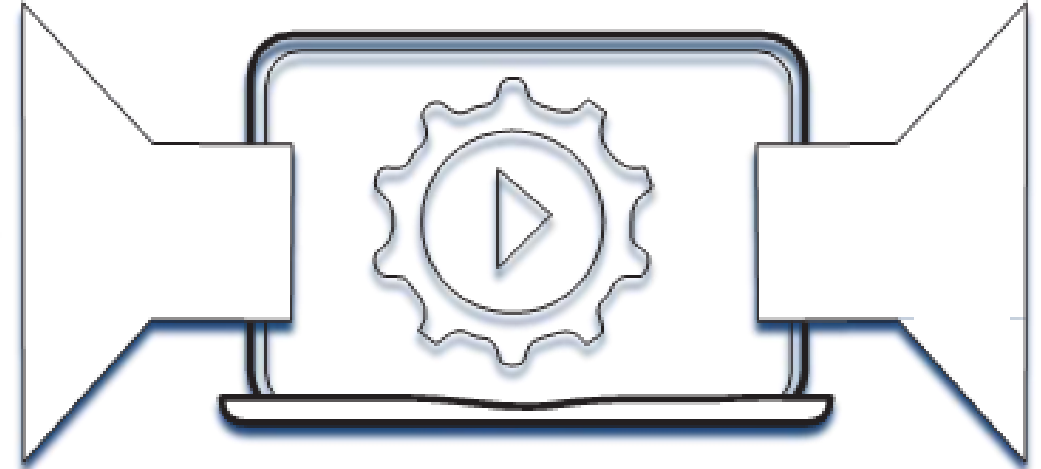
- 47 main questions
- 134 sub-questions
- Numerous sub-sub-questions

Applicability domain of the EDT

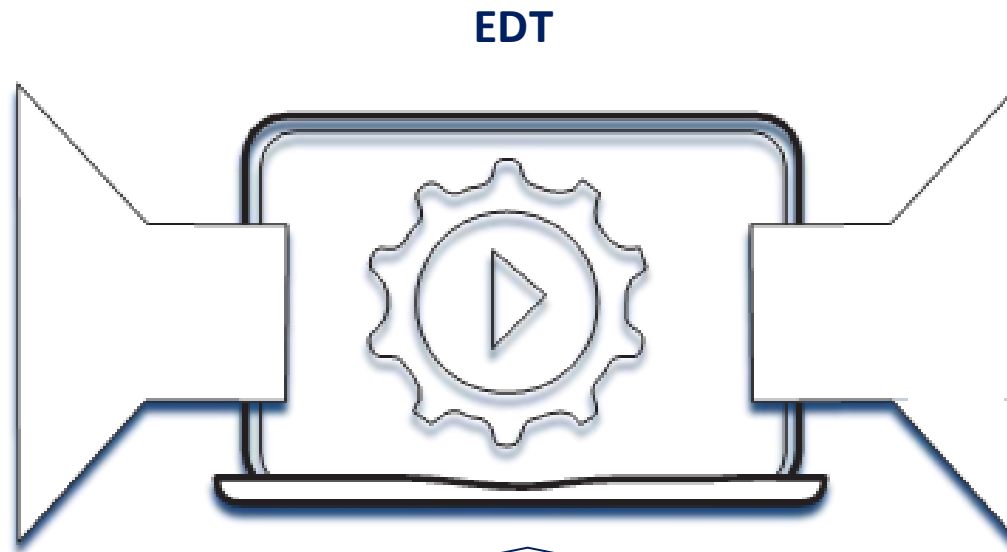
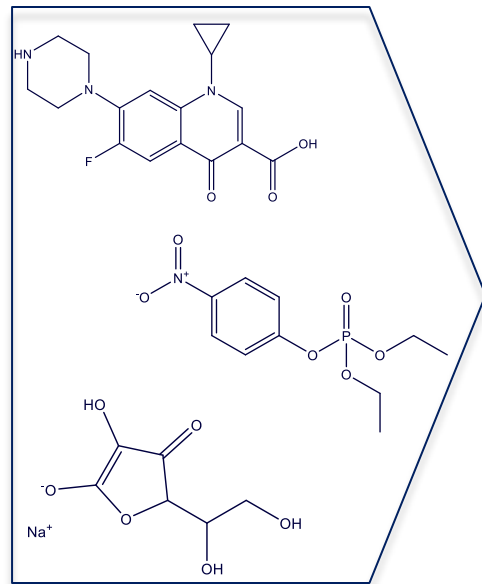
All chemicals **except**:

- Elements and inorganic substances
- Proteins
- Unhydrolyzable polymers

EDT



Six classes of relative chronic oral toxic potential



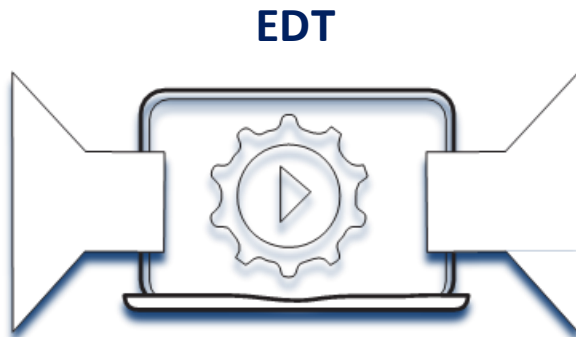
Basis:
the Cramer et al. (1978) DT, a collection of toxicological & ADME* data and info on structure-activity relationships on thousands of compounds

Relative Chronic Oral Toxicity

- Class I
- Class II
- Class III
- Class IV
- Class V
- Class VI

*Absorption, distribution, metabolism, and excretion

EDT TTCs



Relative Toxicity	TTC Level*
Class I	28,400 $\mu\text{g/p/d}$
Class II	4,240 $\mu\text{g/p/d}$
Class III	1,010 $\mu\text{g/p/d}$
Class IV	196 $\mu\text{g/p/d}$
Class V	4.79 $\mu\text{g/p/d}$
Class VI	0.108 $\mu\text{g/p/d}$

* Preliminary values, pending QC

EDT TTCs vs CDT TTCs



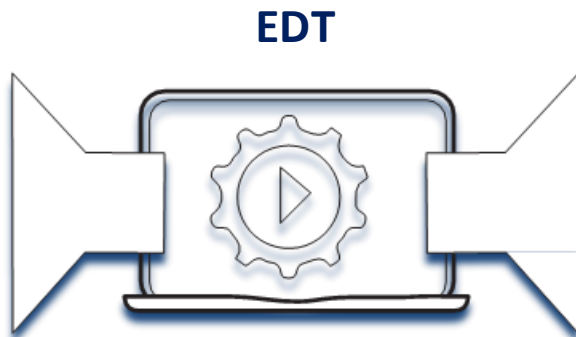
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Class V	4.79 $\mu\text{g/p/d}$
Class VI	0.108 $\mu\text{g/p/d}$

VS

CDT TTC range of 1,800 to 90 $\mu\text{g/p/d}$

* Preliminary values, pending QC

Applications of the EDT



Relative Toxicity	TTC Level*
Class I	28,400 $\mu\text{g/p/d}$
Class II	4,240 $\mu\text{g/p/d}$
Class III	1,010 $\mu\text{g/p/d}$
Class IV	196 $\mu\text{g/p/d}$
Class V	4.79 $\mu\text{g/p/d}$
Class VI	0.108 $\mu\text{g/p/d}$

- Combined with the TTC levels, the EDT may be used for
- ✓ Safety evaluation (tool in the toolbox; acute vs chronic toxic potential)
 - ✓ Screening and prioritization of chemicals for further testing; identify compounds of concern
 - ✓ Post-market: ensure that compounds at current intake levels are still safe
 - ✓ Read-across
 - ✓ Mixture evaluation and determine sufficient similarity of mixtures
 - ✓ May be used to support cumulative toxicity assessment

* Preliminary values, pending QC

Our goals

01

Predictive tox tool

To develop a fast and efficient tool capable of predicting the relative chronic oral toxic potential of chemicals.

02

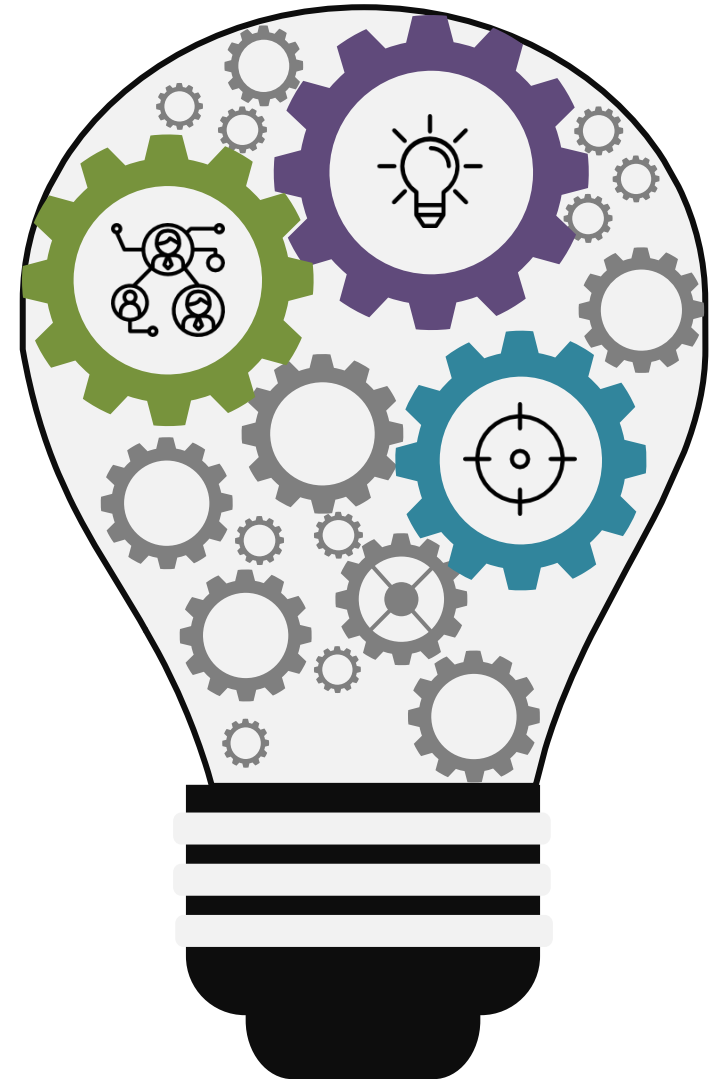
3R

To provide scientists with a tool that can help replace, reduce, and/or refine animal testing.

03

Gain and spread knowledge

To advance our understanding of structure-toxicity relationships and, in general, toxicology.



Other EDT Team Members:

- Timothy Adams, Ph.D. (retired)
- Renata Kolanos, Ph.D.

EDT Software Team:

- Patra Volarath, Ph.D.
- Kirk Arvidson, Ph.D.

Many thanks to:

- CFSAN and OFAS Leadership & Scientists
- All those reviewing the EDT publications
- Drs. Antony Williams and Richard Judson from US EPA

Thank You!



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FDA

U.S. FOOD & DRUG

ADMINISTRATION

CENTER FOR FOOD SAFETY & APPLIED NUTRITION